

WELCOME TO CHEMISTRY
EXPECTATIONS, REQUIREMENTS, GRADING POLICY
MR. RICE, ROOM 206. OFFICE HOURS: LUNCH, AFTER SCHOOL.

A. Requirements

1. **Attendance:** In order for you to do well in this class you must attend class everyday prepared to learn and participate. Our curriculum is cumulative; every chapter builds on the previous chapter. So if you are absent often you will find this course extremely difficult. **Late work is not accepted.** If you have an excused absence or an extracurricular activity you have as many days to turn in the assignment as you missed. All assignments are kept in the box in the back of the room according to date. By district rule, ten or more absences results in an automatic F.
2. **Tardiness:** You need to be in your seats when the bell rings, not walking through the door. The first two tardies will result in a parent contact. Every tardy there after will result in an afterschool or lunch detention. Be punctual please.
3. **Materials for class:** Our Chemistry textbook shall be covered at all times. Fines for damages or loss of book have to be paid, or certain activities such as graduation will not occur. **A three ring binder with dividers will also be required for use as your notebook strictly for chemistry.** This notebook will have three sections. **Section 1:** Signed policy statement and syllabus. **Section 2:** Notes. **Section 3:** All graded assignments. **A scientific calculator is also required. TI-30XIIS recommended.**
4. **Assignments:** An agenda will be kept on the overhead with daily assignments, homework and long term projects to help you stay on track with due dates and dead lines. Homework will be collected at the beginning of class the day it is due.
5. **Lab fee:** A 20 dollar lab fee is required for this class. A check made out to Mcqueen High School or cash paid to our book keeper.

B. Expectations:

1. **Effort: You will be expected to make an effort in this class and earn a good grade.** Chemistry is difficult for some students. You will be able to learn the material covered in this class if you put forth some effort. If at first you don't understand put some more time into it. Ask questions in class. See me at lunch or after school. Every student that has failed this class fails for the same reason, not turning in assignments. Every student in this class can earn an A if they put out the effort.
2. **Students:** You should expect to enjoy, learn and be respectful of others in this class. Please raise your hands for questions and discussions so as to reduce chaos and increase order.
3. **Teachers:** These same expectations I have of you, I also have for myself. I expect to work hard and enjoy this class. I respect each individual's right to learn and participate.

- C. Weighted scale:** Homework = 20%, labs = 20%, tests and quizzes = 45%, final = 15%.

Grades scale: 100-90=A
89-80=B
79-70=C
69-60=D
59-0=F

D. Classroom rules:

1. When the teacher is speaking, do not speak.
2. During group discussions one student speaks at a time.
3. Food and beverages are not allowed in this class including gum.
4. Raise your hand when you have a question or comment.
5. Be in your seats when the bell rings.
6. No hats no headphones no makeup.
7. If I hear any cell phone it will be placed in the cell phone receptacle and your parents can pick it up after school in the student relations office.
8. Respect.

E. I have read and understand the above requirements, expectations and grading policy.

1. Student signature _____
2. Parent or guardian signature _____
3. Date _____

CHEMISTRY SYLLABUS

Fall Semester

Chapter 3: Matter- Properties and Changes

Chapter 4: The Structure of the Atom

Chapter 6: The Periodic Table and Periodic Law

Chapter 7: Ionic Compounds and Metals

Chapter 8: Covalent Bonding

Chapter 9: Chemical Reactions

Chapter 2: Analyzing Data

Chapter 10: The Mole

Spring Semester

Chapter 11: Stoichiometry (Calculations of Quantities in Chemical Reactions)

Chapter 12: States of Matter

Chapter 13: Gases

Chapter 15: Energy and Chemical Change (Thermochemistry)

Chapter 14: Mixtures and Solutions

Chapter 18: Acids and Bases